

Measurement of CO densities by IR absorption spectroscopy in a CO₂/O₂ and CO₂/N₂ dielectric barrier discharge

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The continued and increasing use of fossil fuels, owing to its limited availability, poses a major energy supply and environmental problem. An alternative to this issue is the use of renewable energies. To counterbalance their intermittency, one possible solution is to store the excess energy in a chemical form by reforming CO₂ into a fuel. The most energy-consuming step in this process is the dissociation of CO₂ into CO. A promising route to achieve high conversion, while keeping investment and running costs low, is to use plasma-assisted dissociation [1].

The aim of this work is focused on the fundamental study of CO₂ splitting by a dielectric barrier discharge (DBD) operated at elevated pressure (400 – 700 mbar). Today, the majority of studies on the CO₂ dissociation in DBD have been performed in pure CO₂ [2], while no investigation on the effect of the addition of molecular gases such as O₂ or N₂ in CO₂ DBD at high pressure have been reported, even though it is known that oxygen and nitrogen can influence the CO₂ dissociation [3]. In this study, we will present the influence of the addition of O₂ and N₂ in the gas mixture in a DBD.

The experiments were carried out in a discharge consisting of two pin electrodes covered with glass. The plasma was powered with a sinusoidal high voltage (15 – 22 kV peak to peak) at a frequency of 22 kHz. The electrodes were mounted inside a cylindrical glass cell to maintain reproducible gas conditions. A gas flow at a flow rate in the range of 400-700 standard cm³ per minute (sccm) was established in the cell, while using a vacuum pump at the exhaust. The experiments were performed in pure CO₂ and in different gas mixtures: oxygen/CO₂ and nitrogen/CO₂.

The CO density in the effluent of the DBD was measured *ex-situ* by IR absorption spectroscopy using a pulsed quantum cascade laser, in the range 2212 – 2216 cm⁻¹. The exhaust of the gas was directed into an external cell through which the laser beam passes twice. The total optical path length was 948 mm. The intensity of the transmitted laser light was measured by a fast IR detector and recorded by an oscilloscope.

Influence of O₂/N₂ on conversion efficiency of CO₂ is studied using three different gas compositions. The results are shown in *Figure 1* for different gas flows (square: 400 sccm, circle: 700 sccm) and pressures (square: 400 mbar, circle: 700 mbar). Addition of N₂ increases conversion efficiency while addition of O₂ shows only a dilution effect, acting like an inert spectator.

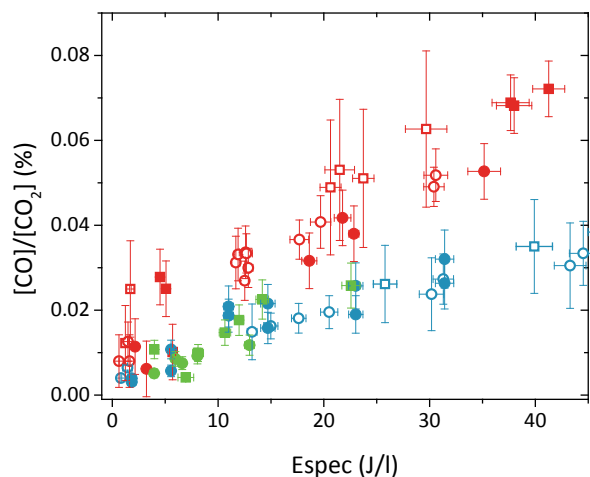


Figure 1: Conversion efficiency in different gas mixtures (blue: 100% CO₂, red 50/50 CO₂/N₂ and green 50/50 CO₂/O₂), pressures (open: 400 mbar, full: 700 mbar) and gas flows (square: 400sccm, circle: 700 sccm) as a function of the specific energy (E_{spec}).

1. References

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